

Periodic Table

1 1.01 1+, 1- H hydrogen							2 4.00 He helium		
3 6.94 1+ Li lithium	4 9.01 2+ Be beryllium			5 10.81 B boron	6 12.01 C carbon	7 14.01 3- N nitrogen	8 16.00 2- O oxygen	9 19.00 1- F fluorine	10 20.18 Ne neon
11 22.99 1+ Na sodium	12 24.31 2+ Mg magnesium			13 26.98 3+ Al aluminum	14 28.09 Si silicon	15 30.97 3- P phosphorous	15 32.07 2- S sulfur	15 35.45 1- Cl chlorine	18 39.95 Ar argon

Legend for Elements

<div style="display: flex; justify-content: space-around; align-items: center;"> <div style="border: 1px solid black; padding: 5px; text-align: center;"> Solid </div> <div style="border: 1px solid black; padding: 5px; text-align: center;"> Gas </div> </div> <p>Note: The legend denotes the states of elements at a temperature of 25°C</p>	<p style="text-align: center;">Key</p> <div style="display: flex; align-items: center;"> <div style="margin-right: 10px;"> Atomic number → </div> <div style="border: 1px solid black; padding: 5px; margin-right: 10px;"> <div style="display: flex; justify-content: space-between;"> 3 6.94 </div> <div style="text-align: center; margin: 5px 0;"> Li </div> <div style="text-align: center;"> lithium </div> </div> <div style="margin-left: 10px;"> ← Atomic molar mass ← Common ion charges (most common first) ← Name </div> </div> <div style="margin-top: 10px;"> Symbol → </div>
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